
Worksheet Mass Problems Answers

? g zncl = 45.0 g al = 341 g zncl 2 27.0 g al 2 mol al 1 ... - a. a mass of 45.0 grams of aluminum will react with how many grams of zinc chloride? ? g zncl 2 = 45.0 g al ! 1 mol al 27.0 g al ! 3 mol z ncl 2 2 mol al ! 136.4 g z 2 1 mol zncl 2 = 341 g zncl 2 b. what mass of aluminum chloride will be produced if 22.6 grams of zinc chloride are used up in the reaction? ? g alcl 3 = 22.6 g zncl 2 ! 1 mol zncl 2 ... **molar mass worksheet answer key - river dell regional ...** - molar mass worksheet ... "molar mass" is used to describe the mass of one mole of a chemical compound, while "atomic mass" is used to describe the mass of one mole of an element or the mass of one atom of an element. 8) which is a better unit for expressing molar mass, "amu" or "grams/mole"? ... **worksheet: mass/mass problems name** - title: microsoft word - 8-20, 21 mass/mass problems wkstc author: brent white created date: 7/13/2005 11:06:30 pm **chemistry computing formula mass worksheet** - find the formula mass of the following compounds. round atomic masses to the tenth of a decimal place. place your final answer in the formula mass column. chemistry computing formula mass worksheet problem set-up example: find the formula mass of ca(no3)2 ca: 1 x 40.1 = 40.1 n: 2 x 14.0 = 28.0 o: 6 x 16.0 = 96.0 ___ **stoichiometry worksheet #1 answers - my chemistry class** - stoichiometry worksheet #1 answers 1. given the following equation: $2c_4h_{10} + 13o_2 \rightarrow 8co_2 + 10h_2o$, show what the following molar ratios should be. a. c_4h_{10} / o_2 b. o_2 / co_2 c. o_2 / h_2o d. c_4h_{10} / h_2o ... what mass of iron is needed to react with 16.0 grams of sulfur? 27.87 g fe b. how many grams of fes are produced? **stoichiometry practice worksheet - social circle city schools** - stoichiometry practice worksheet solve the following stoichiometry grams-grams problems: 1) using the following equation: $2naoh + h_2so_4 \rightarrow 2h_2o + na_2so_4$ how many grams of sodium sulfate will be formed if you start with 200.0 **density practice problem worksheet - chsd** - density practice problem worksheet 1) a block of aluminum occupies a volume of 15.0 ml and weighs 40.5 g. what is its density? 2) mercury metal is poured into a graduated cylinder that holds exactly 22.5 ml. the mercury used to fill the cylinder weighs 306.0 g. from this information, calculate the density of mercury. **worksheet: mole/mass problems name** - a. determine the mass in grams of nh₃ formed from 1.34 moles of nitrogen. b. what is the mass in grams of hydrogen required to react with 1.34 moles of nitrogen? c. how many moles of nitrogen are required to produce 11.7 moles of nh₃? d. how many moles of nitrogen are required to produce 11.7 grams of nh₃? e. **name per: date: density worksheet** - name per: date: density worksheet ... are weighed, the lead would have a greater mass than the feather. it would take a much larger volume ... mass or $d = v m$ to solve density problems, list the known and unknown values, then use one of the following. **molar mass worksheet - nclark** - molar mass worksheet calculate the molar mass of the following chemicals: 1) cl₂ 2) koh 3) becl₂ 4) fecl₃ 5) bf₃ 6) ccl₂f₂ 7) mg(oh)₂ 8) uf₆ 9) so₂ 10) h₃po₄ 11) (nh₄)₂so₄ 12) ch₃cooh 13) pb(no₃)₂ 14) ga₂(so₃)₃ **stoichiometry practice worksheet - hazleton area school ...** - molar mass a mole ratio from molar mass b the balanced equation double lined boxes are conversion factors to convert from one quantity to another. mole ratios reactant s and s y lar mass lar mass y lar mass y y lar mass

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